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Program

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RPGR2012: Realize Challenges, Exchange Ideas The Institute of Physics (IoP), Chinese Academy of Sciences (CAS), Beijing, China. The 4th International Conference of Recent Progress in Graphene Research, Oct. 3- 6, 2012, Beijing

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Structure of Magnetic Clay of Bentonite

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By sequential deposition of Fe2+ and Fe3+ in the presence of ammonia and bentonite obtained magnetic clay. In the magnetic composite detected peaks corresponding aluminum-oxygen and silicon-oxygen groups of montmorillonite and magnetite by method of X-ray diffraction. Shown that the size of magnetite particles in the structure of clay is 10-20 nm.

One of the important problems of modern science is to obtain materials with desired properties. In this regard, of particular interest to obtain and study the magnetic properties of sorbents. The structure of magnetic clay obtained by sequential deposition of salts Fe (II) and Fe (III) in the presence of bentonite and ammonia was studied. The diffraction pattern of bentonite clay found two peaks in the 2θ =6,80 and 20,10, corresponding silicon oxide and aluminum-oxygen groups in the structure of montmorillonite - the main component of bentonite. In the spectrum of magnetite also found two peaks at 2θ =30,60 and 37.20, corresponding to γ -Fe2O3 - bematite and FeO•Fe2O3 (Fe3O4) - magnetite.

X-ray analysis of magnetite – clay's system shows a peaks corresponding to the components of benconte and magnetite. However, the intensity of the peaks was changed. As the concentration of magnetite decreases the intensity of the peaks corresponding to silicon oxide and aluminum-oxygen groups and increases the intensity of the peaks corresponding to the component of magnetite.

Electron microscopic studies of the magnetic clay showed the presence in the structure of bentonite dark costs size of 10-15 nm, corresponding to particles of magnetite. Such particles can be in inter-packet space of montmorillonite. With increasing content of magnetite in the structure of clay until 6.8% increase in the size of magnetite occurs in the structure of clay, which is due to the saturation of cation exchange capacity. The maximum size of the magnetic particles is 20 nm, this corresponds to the size of inter-packet space of montmorillonite.

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