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## nano-sized films embedded with silver and titanium Electrical properties and structure of amorphous diamond-like carbon

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films practical use [1,2]. films leads to the appearance of their new physical properties and expands the possibilities of the Embedding of metal with different chemical nature into amorphous diamond-like carbon (a-C:H)

that the a-C:H<Ag+Ti> films are more thermally stable. At the same time, electrical properties and the structure of the films have not been studied. absorption spectrum in a-C:H<Ag+Ti> films, as well as in the a-C:H<Ag> films [3], and it was found It was observed clear pronounced optical plasmon resonance in the visible region in optical

a-C:H<Ag+TI> films with thickness from 80 to 100 nm were deposited on quartz and silicon (100) at. %, respectively, and was changed by alteration of metals and graphite area ratio on the target substrates at 50°C. The maximum of Ag and Ti metals concentration in the films reached 4,2 and 20 prepared by ion-plasma DC magnetron sputtering of the combined target in Ar+CH<sub>4</sub> gas mixture. The embedded with silver and titanium (a-C:H<Ag+Ti> films) are presented in this work. The films were Results of electrical properties and structure study of amorphous diamond-like carbon films

a-C:H<Ag> films. Temperature dependences of the a-C:H <Ag+Ti> films conductivity, as well as that the concentration dependence of conductivity in the films is less pronounced in comparison with the magnitude) with Ag concentration rise at fixed Ti concentration. Note that percolation character of of the a-C:H<Ag> films, is typical of semiconductors. It was found the significant increase in the a-C:H<Ag+Ti> films conductivity (  $\sim$  10 orders of

2-3 nm. Changes in the electrical properties of the a-C:H<Ag+Ti> films are apparently due to the isolated nanoparticles of silver (~ 60 nm), and those of silver and titanium oxide with dimensions of Study of the a-C:H<Ag+Ti> films structure by the TEM showed that their matrix contains

sp<sup>3</sup>/sp<sup>4</sup> ratio change of the hybridized bonds in their matrix. electrical properties with preserve of plasma absorption resonance. Thus, embedding of Ag and Ti into the a-C:H films could lead to effective modification of their

## References

1. R. Videnovic, V. Thomson, P. Oelhafen, Applied physics letters (2002) 80, 2863

2. Jinfeng, Q. Li and Zh. Bin et. al., Applied Surface Science (2012) 258, 5025

3. Yu. Prikhodko, S.L. Mikhailova, Ye.S. Mukhametkarimov, et. al, Proc. of SPIE (2016) 9929 9929G-1.

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Carbon Nanostructures

## Structural and electronic properties of nanostructured graphite

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energy calculations within DFT and pseudopotential method as implemented in [1]. configuration corresponds to the minimum of total energy We have performed self-consistent total within the slab scheme. The nearest to the graphene  $C_{00}$  pentagon was centered on carbon atom. This Hybrid system of graphene-C60 was used for calculations. We considered superlattice of 116 atoms Ab initio simulation of nanostructured graphite was performed using density functional theory

have been determined means of structural relaxation (Fig.1). The electronic energy spectrum of graphene-C<sub>60</sub> hybrid system convergence was at least 10° Ry. Local atomic structure of graphene-C<sub>60</sub> hybrid system was studied by the Monkhorst-Pack method with a two-dimensional 4x4x1 mesh. The achieved total energy cutoff energy was 200 eV. For calculation of all surfaces, we used the k-points generation scheme by was also studied. The effective charges on the nearest neighbor atoms of the fullerene and graphene approximation (GGA). For the plain waves used in the expansion of the pseudo-wavefunctions, the For the exchange-correlation energy we used the PBE functional within the generalized gradient



References Fig.1. Supercell model of graphene-C<sub>m</sub> hybrid system after structural relaxation (a - side and b - top views)

1. P. Giannozzi et al. J. Phys.: Condens. Matter. (2009) 21, 395502