

PROGRAM SCHEDULE



**CARBON 2019**  
LEXINGTON, KY



JULY 14 - 19

- 20** Structural integrity of polyelectrolyte/graphene oxide layer-by-layer assemblies in concentrated aqueous solutions **Szabo**
- 50** The synthesis of micro and nanodiamonds by oxygen-acetylene torch method **Partizan**
- 72** SERS spectra of rhodamine using Ag nanoparticles photodeposited onto graphene oxide thin films **Costa**
- 112** The role of carbon nanotube-graphene hybrid structure for high performance LED **Seo**
- 116** Electrochemically Desulfurized Molybdenum Disulfide (MoS<sub>2</sub>) - Graphene Aerogel Composites As Highly-Efficient Electrocatalysts for Hydrogen Production **Gupta**
- 117** Novel Fabrication for Graphene Oxide/Polymer-based Thin Film Composite Membranes with Enhanced Filtration and Antifouling Characteristics **Gupta**
- 118** Graphene-mediated Surface Enhanced Raman Spectroscopy for Detection of Biomolecules and DNA Hybridization **Gupta**
- 151** The impact of sonication on graphene oxide's physicochemical properties and nanofiltration performance of covalently crosslinked membranes **Kandjou**
- 180** Flexible Microsupercapacitors with Flash Lamp Processed Graphene-Carbon Nanotube Electrodes **Kim**
- 237.1** 1 electron paramagnetic resonance in graphene structures **Baitimbetova**
- 243** Graphene reactions with water and carbon dioxide on armchair sites **Oyarzún**
- 359** Hybrid Graphene oxide-Zirconium oxocluster polymer for fluoride removal: Synthesis, characterization and water treatment. **Pérez-Tavares**
- 370** Molecular-Scale Graphene Oxidation and Pitting Dynamics **Schmitt**
- 384** Electronic and magnetic properties of graphene oxide /Fe<sub>2</sub>O<sub>3</sub> nanocomposites **Idisi**
- 396** Advanced Bioinspired Membranes from Biological Ion Channel Confined into Polymeric Cylindrical Nanopores for Enhanced Filtration **Gupta**
- 406** HREELS measurements of the phonon dispersion for graphene **Singh**
- 460** DFT study on the conversion of methane to benzene during CVI. **Li**
- 479** Challenges in Graphene Nanopore- (Nanowindow)-based Molecular Separations: A Critical Review **Vallejos-Burgos**
- 490** Method of graphene synthesis in the combined flame **Lesbayev**
- 513** Nitrogen Doped Functionalized Graphene as Sustainable Photocatalyst for Visible Light Induced Overall Water Splitting **Tripathi**
- 526** Green exfoliation of graphene nanosheets based on freezing induced volumetric expansion of carbonated water **Kaur**
- 527** Nature-inspired, Graphene-wrapped 3D MoS<sub>2</sub> Ultrathin Microflower Architecture as a High-Performance Anode Material for Sodium-Ion Batteries **Anwer**
- 533** Electrical properties of MOFs/GO composites **Devautour-Vinot**
- 537** Chemically Hydrogenated Graphene for Solid-State Hydrogen Storage Applications **Morse**
- 540** Low Energy Electron Transparency of Graphene and Doped-Graphene via Electron Energy Analyzer **Belayneh**

- 14** Effect of Filler Size in a Graphene-Anthracene Carbon-Carbon Composite **Vander Wal**
- 28** Creation of nano- and micro- spaces in carbons and loading dissimilar materials by electrospinning **Oshida**
- 65** Synthesis of novel nanocomposite of activated Kaolinite with various proportions of modified carbon nanotube for improved adsorption of heavy metals at very low concentration from water **Yadav**
- 74** Electrospun Lignin-based Carbon Nanofiber Tows **Yi**
- 114** Engineered Nanocomposite Materials Properties through Embedding of Smaller Nanoparticles in a Polymer Matrix **Gupta**
- 143** Preparation of Carbon Nanofiber Using Chlorinated PVC/Isotropic Pitch Mixed Solutions by Electrospinning **Ryu**
- 165** Influence of carbon nanotube fillers on change of mechanical properties of polyimide composite films **Yar-Mukhamedova**

## METHOD OF GRAPHENE SYNTHESIS IN THE COMBINED FLAME

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### Introduction

Since the discovery of the first method of obtaining graphene by mechanical splitting of graphite layers, the efforts of many researcher have been aimed at developing more effective approaches to solving actual problem related to the development of a reproducible method for the synthesis of graphene in macroscopic amounts. The flame is an ideal reactor for the production of carbon nanomaterials by the method of assembling them using a bottom-up mechanism, since in the flame, the formation of the final product occurs through successive elementary acts consisting of innumerable of chemical reactions that occur in a very short period of time. At present, the only problem remains the solution of the problem of controlling the reaction route of chemical reactions for the formation of desired combustion products. In the proposed study, a method for the selective use of intermediate combustion products as a building material for the formation of graphene has been developed to solve this problem. The main novelty of this research is to the use of the phenomenon combining of the reaction zones of flames during the combined combustion of different fuels. This allows you to influence the structure and property of the resulting final products of combustion, by changing the composition of intermediate particles in the reaction zone of the combined flame acting as building materials. The main advantages of the proposed method are the synthesis of graphenes in an open atmosphere, the short time of the graphene formation process and the absence of additional energy costs.

### Materials and Methods

In order to create the combustion method that providing the combination of combustion zones of gaseous fuels an experimental setup was designed and created. The major unit of setup for combustion of gaseous fuels is a burner with coaxial arrangement of nozzles. The construction of setup allows to apply the separate fuel to each nozzle individually. The burner provides the possibility of displacement the nozzles that relative to each other on vertical axis, and this fact regulate the concentration density and composition of intermediate particles in combined zone of flames. Obtained samples were examined with the help of transmission (JEM 1011), scanning (Quanta 3D200i) electron microscope, Raman spectroscopy (NTEGRA Spectra Raman,  $\lambda = 473$  nm, the area signal with a diameter of 80 nm).

## Results and Discussion

Research on the formation of graphene layers were carried out in alternation flame of propane with ethanol (propane is fed to the middle) and ethanol with propane (ethanol is fed to the middle). It was found that at the substrate there is formed 5-10 layers of graphene ( $I_G/I_{2D} = 1,64-2,05$ ), Raman spectra and Raman map of which are shown in Figure 1.

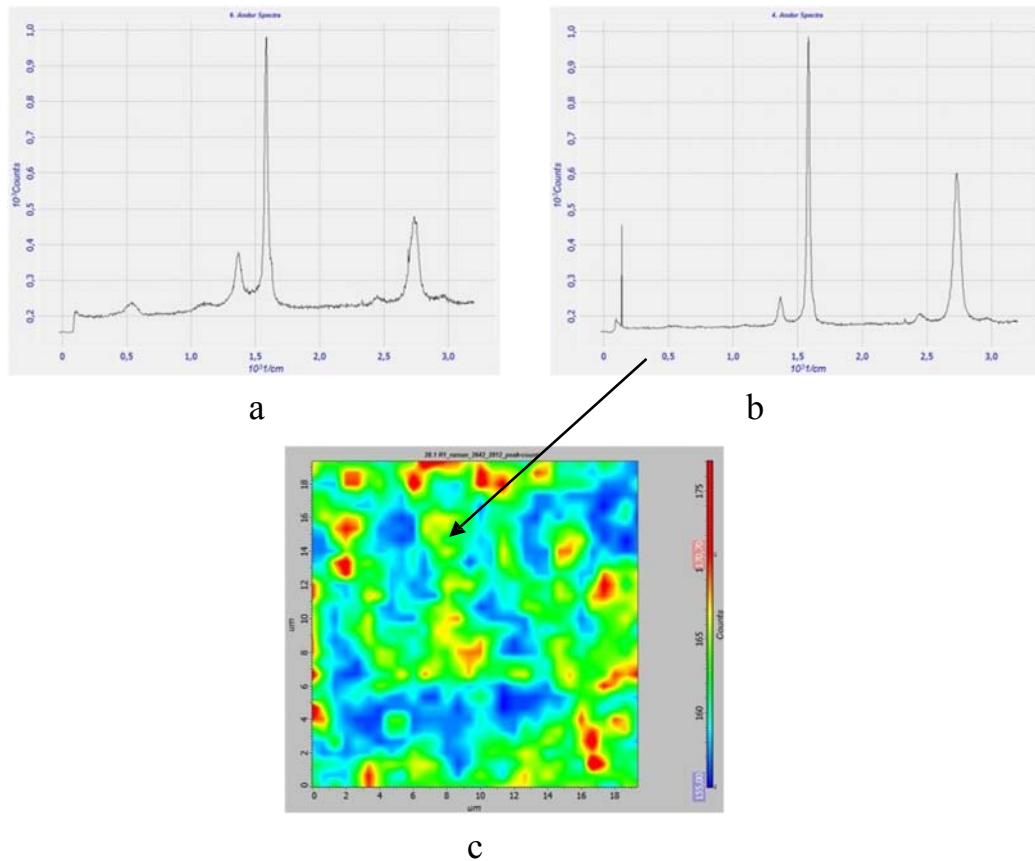


Figure 1 – Raman spectra (a,b) and Raman map (c).

## Conclusions

Studies have shown the possibility of producing the graphene layers in alternation flame of propane with ethanol at atmospheric pressure at nickel substrate with small defectiveness in the range of  $I_D/I_G = 0,26 - 0,39$ . It was found that at fed of ethanol to the center of flame there is formed 5 graphene layers with defective  $I_D/I_G = 0,26$ .

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