



# INESS

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### ABSTRACT BOOK



#### The 4<sup>th</sup> Workshop on Water and Soil Clean-up from Mixed Contaminants



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82.	Kenzhina I. E., Abdullin Kh., Chikhnay Ye. V., Gabdullin M. T., Ismailov D. V. <b>Synthesis of Carbon Composites and Nanocatalysts by Electrospinning Method</b>	92
83.	<u>Khairullina E.</u> , Panov M., Safonov S., Logunov L., Kochemirovsky V. <b>Non-Enzymatic Glucose and Hydrogen Peroxide Sensors Based on Metal Structures Produced by Laser-Induced Deposition from Solution.</b>	
84.	<u>Kim E. R.</u> , Keldinova A. B., Gritsenko L. V., Abdullin Kh. A., Kurnekov S. E. <b>Thermal Treatment of Aluminum Doped Zinc Oxide Thin Films</b>	93
85.	<u>Kim E. R.</u> , Ualikhanov R. E., Ucubassova D. T., Gritsenko L. V., Guseinov N. R., Abdullin Kh. A. <b>Photosensitivity of Nanostructured CdS Layers, Synthesized by Hydrothermal Route</b>	94
86.	<u>Kozlovskiy A.</u> , Kaikanov M., Tikhonov A., Ponomarev D. <b>Synthesis and Modification of Ni-nanotubes by Electron Radiation</b>	95
87.	<u>Nurdillayeva R. N.</u> , Horrocks B. R. and Pike A. R. <b>Electronic and Electrochemical Characterization of DNA - templated CdS nanowires</b>	96
88.	<u>Omirbekov D. B.</u> , Zhunisbekov A. T., Ramazanov T. S., Orazbayev S. A., Dosbolayev M. K., Gabdullin M. T., Zhurnadyllov R. E. <b>Obtaining of Carbon Nanoparticles in Combined RF/DC Discharge Plasma</b>	97
89.	<u>Puzikova D. S.</u> , Khussurova G. M. <b>The Influence of Counterelectrode Material on Photocurrent Generation in Polysulfide Electrolyte</b>	98
90.	<u>Puzikova D. S.</u> , Dergacheva M. B., Khussurova G. M. <b>Photoelectrochemical Cell with Modified CdSe Photoanodes</b>	99
91.	<u>Ryaguzov A. P.</u> , Nemkayeva R. R., Guseinov N. R. <b>Effect of Sn Nanoparticles on Optical Properties of HDLC Films</b>	100
92.	<u>Sarsenov A. M.</u> , Myrzakhmet M. K., Baitassova Zh. Y. <b>New Express Method for the Activity of Solvent Molecules Determining in Electrolyte Solutions</b>	101
93.	<u>Sarsenov A. M.</u> , Sarsenova M. A., Myrzakhmet M. K., Sataeva G. E., Bekmukhanbetova D. B., Baitassova Zh. Y. <b>Dosing of Nano-Microcomponents of Hardly Soluble Substances in Water</b>	102
94.	<u>Serikbaev B. A.</u> , Kamysbaev D. Kh., Arbuz G. S. <b>Sorbents Based on Rice Husk for the Synthesis of Modified Electrodes</b>	103
95.	<u>Shalabayev Zh. S.</u> , Madikassimova M. S., Tatykayev B. B., Uralbekov B. M., Burkibayev M. M., Urakaev F. Kh. <b>Synthesis of CuS/S Nanocomposites and Their Application Fields</b>	104
96.	<u>Sherahan A.</u> , Belgibayeva D. S., Nurpeisova D. T., Smagulova A. <b>Size-Controlled Synthesis of Iron Nanoparticles in Aprotic Polar Solvents</b>	105
97.	<u>Shomanov A. S.</u> , Yessenbayev Z. A., Matkarimov B. T., Beketayev K. B. <b>Design and Simulation of a New Coaxial Probe for NSOM</b>	106
98.	<u>Talamona D.</u> and Tan K. H. <b>Green Concrete using Recycled Aggregate Concrete for Sustainable Construction</b>	107
99.	<u>Tatykayev B. B.</u> , Shalabayev Zh. S., Uralbekov B. M., Burkibayev M. M., Urakaev F. Kh. <b>Photocatalytic Activity of Solid-State Synthesized Associate Nanocrystalline AgCl@Ag</b>	108
100.	<u>Urakaev F. Kh.</u> , Abuyaeva B. B., Vorobyeva N. A., Mun G. A., Uralbekov B. M., Zharlykasivova D. N., Burkibayev M. M. <b>Nanosulfur in the Water-Soluble Polymers: Synthesis and Application</b>	109
101.	<u>Urazov K. A.</u> , Zaretskaya E. P., Gremenok V. F. <b>Electrodeposition-Annealing Process for Preparation Cu<sub>x</sub>ZnSn(S<sub>x</sub>Se<sub>1-x</sub>)<sub>4</sub> Thin Films</b>	110
102.	<u>Urazov K.</u> , Dergacheva M., Gremenok V., Stanchik A., Bashkirov S. <b>Photocharacteristics of Electrodeposited CZTSe Thin Films on Different Substrates</b>	111
103.	<u>Batrushov D. G.</u> , Yerlanuly Ye., Ramazanov T. S., Dosbolayev M. K., Gabdullin M. T. <b>Obtaining of Carbon Nanowalls by PECVD Method in the Plasma of Radio-Frequency Discharge</b>	112
104.	<u>Yermekova A.</u> , Tulebaeva D. <b>Synthesis of Nanostructures Based on Ferrum Oxide</b>	113
105.	<u>Kudyarova Zh. B.</u> , Yerpaiyz B.D., Mironenko A. V., Kazieva B. A., Mansurov Z. A. <b>Catalysts for Dimethyl Ether Synthesis</b>	114
106.	<u>Zhakiyev N.</u> , Sovetov M., Otarov R., Kopanos G. <b>Techno-Economic Analysis of Renewable Power Plant Expansion Considering Provision of Curtailment from Combined Heat and Power (CHP) Plant and Electricity Storage</b>	115
	4th Workshop on Water and Soil Clean-up from Mixed Contaminants	116
107.	<u>Václavíková M.</u> <b>WaSClean - Water and soil clean-up from mixed contaminants</b>	118





## Synthesis of CuS/S Nanocomposites and Their Application Fields

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Copper sulfide and sulfur nanoparticles have many application fields in industry and technology<sup>1, 2</sup>. Nowadays, copper sulfide and sulfur nanocomposites can be used as a semiconductor in electrochemistry, photocatalyst in catalysis and in battery technology<sup>2, 3</sup>. In this report, nanocomposites of CuS@S were co-precipitated via the reaction  $\text{Na}_2\text{S}_n + \text{CuSO}_4 \rightarrow \text{CuS} + \text{S}_8 + \text{Na}_2\text{SO}_4$ . Obtained black precipitate was separated, washed and dried for the further analyses. X-ray powder diffraction analysis (D8 ADVANCE, Bruker) was used for characterization morphology and composition of obtained nanocomposite. X-ray data showed that the sample contained two phases: CuS and S<sub>8</sub> (orthorhombic, Space Group Fddd, no. 70), see Figure 1. According to the semi-quantitative analysis, chemical composition of sample was 67.9% CuS and 32.0% S<sub>8</sub>. Copper sulphide had a hexagonal crystal lattice and sulfur was well crystallized. The crystallite size of CuS and S<sub>8</sub> were 22.4 nm and 54.6 nm respectively.

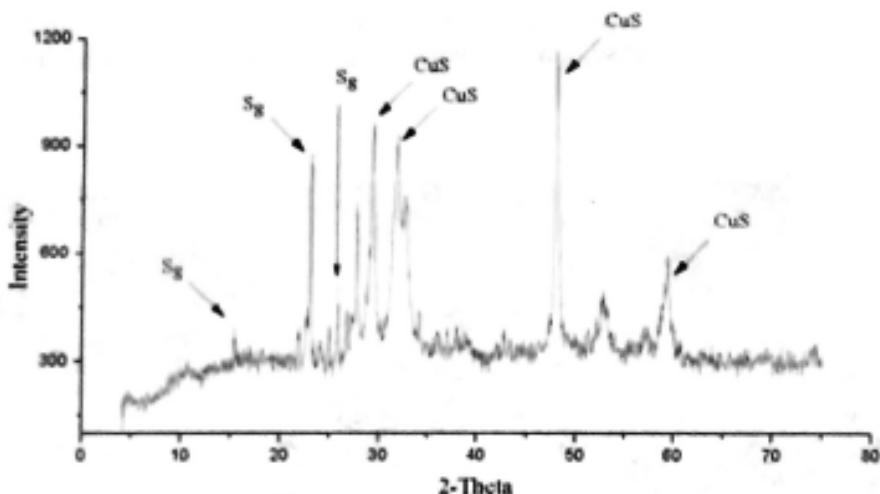


Fig. 1. XRD pattern of CuS/S nanocomposite.

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Katsuhiko T.	49	Mukhanov Anuarbek	50
Kaydar B.	20	Myronov Maksym	19, 69
Kazieva B.A	115	Nakamura H.	43
Keldinova. A.B	94	Nazhipkyzy M	71
Kenzhegaliyeva Elzira	63	Nemkayeva R.R.	101
Kenzhina I.E	91, 92	Noyanbayev Nurym	36
Kerimray Aiymgul	47	Nugent Liam	124
Khairullina E.	93	Nuraje Nurxat	53
Khokkin.A. L	28, 74,	Nurakhmetov D.	44
Khussurova G.M	99, 100	Nurdillayeva Raushan	97
Kim E.R.	94, 95	Nurgain A	71
Kiselev E. A.	64	Nurpeissova D.T	106
Kochemirovsky Vladimir	93	Nurpeissova A.	32, 35, 56, 58, 69, 70, 73
Kolotygin V. A.	54	Okube Maki	39
Kopanos Giorgos	116	Omirbekov D.B	98
Koshkina A. A.	28, 64	Orazbayev S.A.	98
Kosova Nina V.	16, 26, 75	Otarov Rustam	41
Kotuč Juraj	128	Otarov Rustam	116
Kozlovskiy A.	85, 96	Pannala Ananth S.	123
Krtík Petr	39	Panov Maxim	93
Krulakova Maria	129	Pavlenko, V.	34, 89
Krupa Vitazoslav	129	Pelegov D. V.	28, 30, 64, 74
Kudreeva L	88	PEPONIDOU Evgenia	120
Kudreyeva. L	89	Pereira-Ramos J-P.	12
Kudyarova Zh.B	115	Petrykin V.	39, 43
Kulametov Zhalgas	65, 66, 68	Podgornova O	75
Kumekov.S.E.	94	Ponomarev D.	96
Kupka Daniel	119, 121	Poulopoulos Stavros	47
Kurbatov A	34, 81, 84, 89	Presniakov Igor A.	33
Kurmanbayeva Indira	66, 67, 68	Prikhodko N.	34, 71
Kuznetsov.D. K.	28, 74,	PryakhinaV. I.	28, 64, 74
Labus Milan	129	Puzikova D	99, 100
Lazarova Edita	129	Radchik Alex	25
Lee S.	43	Rakhimzhanova. M	61
Lesbayev B	71	Ramazanov T.S.	98, 113
Liam Nugent	124	Rezepova Daria	26
Lim J.S.H.	49	Romanuyk K. N.	30
Litasov Konstantin	77	Ryaguzov A.P.	101
Logunov Lev	93	Rysbekova A	80
Lu Maxim	42	Sadykova. A	66
Ma Xiaodi	24	Safonov Sergey	93
Macounva Katerina	39	Sanbayeva	62, 126, 127
Makhmutov. A. R.	28, 74	(Muzdubayeva) Ayana	
Kunelbayev M.	130	Sandeman Susan	124
Mansurov Z.A.	16, 20, 34, 115	Sarbassov Yerbol	47
Matkarimov B.T.	107	Sarker Dipak K.	123
Mentabayeva Almagul	57, 58, 59, 60, 65, 68,	Savina I.	125
122		Sazonov. R	61
Mengaliyev Duisen	51	Seitov A	55
Mikhailovska L.	125	Sekula Peter	122, 128
Mikhailovsky Sergey	123, 124, 125	Sekula Peter	121, 122, 128
Mironenko A.V	115	Sekula Peter jr	121
Molkanova A.	27, 29, 35, 59, 61, 62, 63,	Sergiienko S.A.	54
69, 73		Serikbaev Bazarbay	90, 104
Mosej Juraj	121, 122	Shabdan Yerkin	53
Mukanova Aliya	35, 69, 73	Shalabayev Zh.S	105, 109
Mukanova Zhansaya	65		
Mukhambetov D.G	85		
Mun G.A.	109		
Munakata Hirokazu	11, 13		